

EFFECT OF LACTIC ACID ON *ENTODINIUM CAUDATUM* MONOCULTURE

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Abstract

This experiment was carried out to evaluate the effect of lactic acid on *Entodinium caudatum* monoculture in vitro. After thawing, *E. caudatum* was grown at 39°C under anaerobic condition to yield 10⁵. Four groups were established by inclusion of 0, 0.5, 1, and 2 mM DL-lactic acid (Fluka Chemicals, 69775). *E. caudatum* started to selectively use lactate to maintain 1.2 mM concentration at the highest lactic acid concentration. Increasing lactic acid concentration in medium was associated with reduction in pH ($P < 0.0001$) and increase in total volatile fatty acids ($P < 0.0001$), but no change in ammonia concentration. There was a reduction in acetate ($P < 0.04$) and increases in propionate ($P < 0.02$) and butyrate ($P < 0.0001$) proportions as lactic acid concentration in medium increased. Stoichiometrically calculated gas production and CH₄ amount increased accordingly with total volatile fatty acid production. In conclusion, *E. caudatum* grows to utilize lactate in case of acidosis.

Keywords: *Entodinium caudatum*, lactic acid, in vitro rumen.

INTRODUCTION

In order to meet energy demand, feeding excessive amount of readily fermentable carbohydrate sources can disturb rumen flora and microbial fermentation, which may result in acute and/or subacute acidosis (Umucalılar and Gülşen, 2005; Umucalılar et al., 2012).

Protozoa, especially ciliates play a significant role in lactic acid metabolism in rumen when excess grains are fed. They engulf starch and soluble carbohydrates, which limits their utilization by amylolytic bacteria. This reduces lactic acid production (Nagaraja et al., 1992). Moreover, protozoa increases lactate fermentation, which reduces lactic acid accumulation in rumen (Nagaraja et al., 1992; Russell and Hespell, 1981). Entodiniomorphid ciliates help maintain ruminal pH (Dehority, 2005), by storing starch to minimize its utilization by starch-utilizing bacteria (Schwartzkopf-Genswein et al., 2003). This in vitro experiment was conducted to evaluate *E. caudatum* cultures in response to increased lactic acid concentration in medium.

MATERIALS AND METHODS

After thawing frozen *E. caudatum* cultures at 39°C, they were allowed to grow in Medium M at 39°C under anaerobic conditions to enumerate 10⁵ (Dehority, 1998). Media were enriched 1.5% wheat flour and 1% ground alfalfa daily.

Cultures were then added with 0, 0.5, 1, and 2 mM DL-lactic acid (Fluka Chemicals, 69775). Upon condensation, 1 of 10th of the sediment were added with 96.6 ml Medium M, to achieve 10³-10⁴/ml. After incubation at 39°C, 0.2 ml medium and 1.2 ml substrate solution were refreshed everyday at the same time. Every 3 d, half of the media was added with fresh Medium M (Dehority 1998). Media pH were measured before adding and 5 h after adding the substrate solution. On d 5 and refreshment of the media 1 ml sample was taken for determination of lactic acid and volatile fatty acid (VFA) concentrations and enumeration of protozoon.

Stoichiometrical Calculations (Blümmel et al., 1999):
CO₂ production (CO₂fer), mmol = acetate/2 + propionate/4 + 1.5 x butyrate

CH₄ production (CH₄fer), mmol = acetate + 2 x butyrate - CO₂
 CO₂ released from buffer (CO₂buff), mmol = total VFA
 Gas production, ml = (CO₂fer + CH₄fer + CO₂isobutyrate + CO₂buff) x 0.0821 x 312
 Methane level (Wolin, 1960; Ramin and Huntanen, 2012) was calculated using formula as follows:
 Methane, ml = 22.4 x [(0.5 x acetate) - 0.25 x propionate) - (0.5 x butyrate) - (0.25 x valerate)]
 In a completely randomized design experiment data were analyzed using 2-way ANOVA (SPSS, 2006).

RESULTS AND DISCUSSIONS

Increasing lactic acid addition up to 2 mM decreased pH (Table 1; Figure 1). Excess lactate appeared to be used by *Entodinium*

caudatum to maintain its level by 1.2 mM (Figure 2).

Table 1. Effects of lactic acid addition on medium pH, ammonia, lactate concentrations and *Entodinium caudatum* numbers

Trt ¹	pH	Ammonia mM	Lactate mM	Protozoon
0	6.60	2.98	---	4.7x10 ³
0.5	6.61	2.91	0.83	6.6x10 ³
1	6.56	3.02	1.17	8.1x10 ³
2	6.50	3.04	1.27	9.4x10 ³
Effect	P > F			
Trt	0.0001	0.31	0.0001	0.0001
T	0.0001	0.0001	0.0001	0.0001
Trt x T	0.001	0.08	0.0001	0.0001

¹Trt = treatments, lactic acid, mM. T = time, day.

Table 2 Effects of lactic acid addition to medium containing *Entodinium caudatum* monocultures on VFA profile and fermentation parameters

Parameters	Lactic Acid (mM)				SEM	P > F
	0	0.5	1.0	2		
Acetate (%)	56.4	57.8	54.9	52.8	0.84	0.040
Propionate (%)	22.2	21.5	23.6	24.3	0.44	0.017
Isobutyrate (%)	6.0	5.5	5.4	5.0	0.13	0.009
Butyrate (%)	9.1	9.1	10.1	12.5	0.32	0.000
Isovalerate (%)	4.9	4.8	4.5	4.0	0.27	0.572
Valerate (%)	1.4	1.2	1.5	1.4	0.07	0.373
Σ VFA (mM)	0.42	0.46	0.45	0.52	0.01	0.001
CO ₂ fer (ml)	0.20	0.22	0.22	0.27	0.01	0.000
CH ₄ fer (ml)	0.11	0.13	0.12	0.13	0.001	0.028
CO ₂ buff (ml)	0.40	0.43	0.42	0.49	0.01	0.000
Gas (ml)	18.8	20.4	20.0	23.6	0.45	0.000
NGR ¹	3.60	3.82	3.36	3.36	0.09	0.070
e-CH ₄ ² (ml)	2.52	2.80	2.61	2.97	0.06	0.027

¹NGR = nonglucogenic VFA:glucogenic VFA.

²e-CH₄ = estimated methane production.

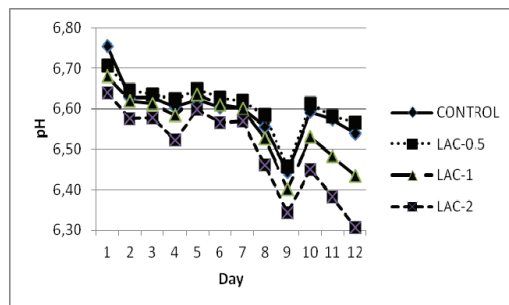


Figure 1. Alterations in pH in media containing *E. caudatum* upon addition of different concentrations of lactic acid (SEM = 0.06).

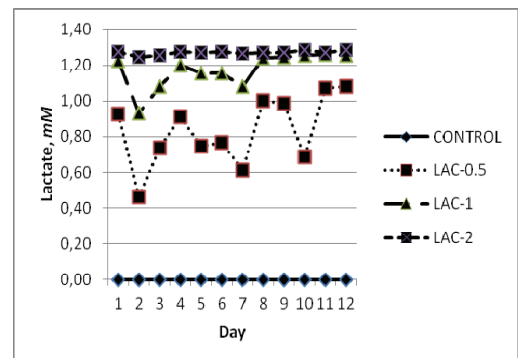


Figure 2. Alterations in lactate concentrations in media containing *E. caudatum* upon addition of different concentrations of lactic acid (SEM = 0.01).

While total VFA increased, proportion of acetate decreased and proportions of propionate and butyrate increased as concentration of lactic acid increased in media containing *Entodinium caudatum* (Table 2).

Increased lactic acid concentration caused increases in total VFA production and CO₂ as well as CH₄. Increased CO₂ release from buffer is a way to neutralize pH. These increases led to increases in stoichiometrically calculated gas production and CH₄ (Table 2).

CONCLUSIONS

Lactic acid inclusion up to 2 mM decreased pH in media containing *E. caudatum*. Reduction in pH associated with stimulation of *E. caudatum* to maintain pH, through modifying rumen fermentation.

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